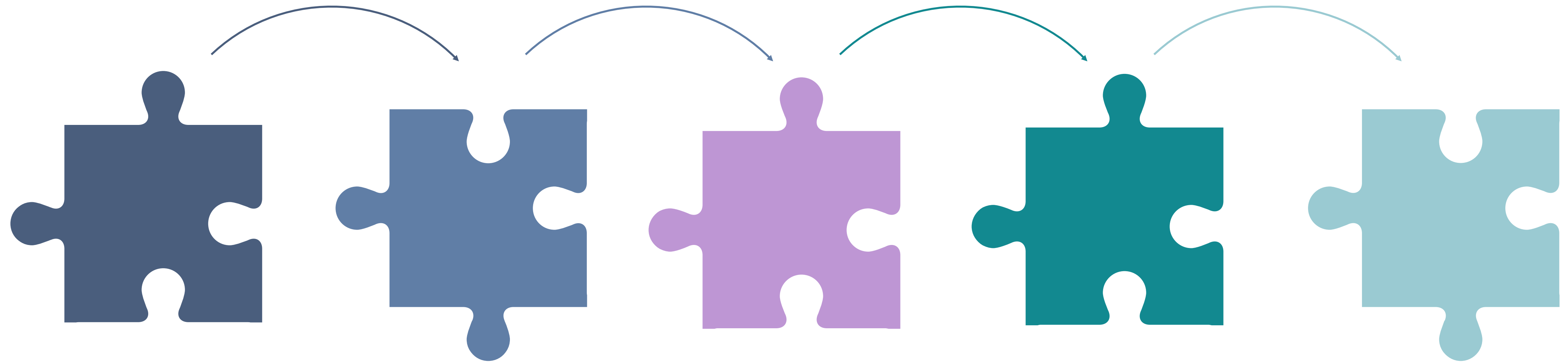




new tool for precise
quantification of pesticide
degradation in
contaminated aquifers

IsoFLUX





1. Preferential pathways

Are there preferential pathways?
If so, where are they located?

2. Contaminant mass

How much contamination is migrating? Is this a relevant mass to be considered a migration risk?

3. Migration rate

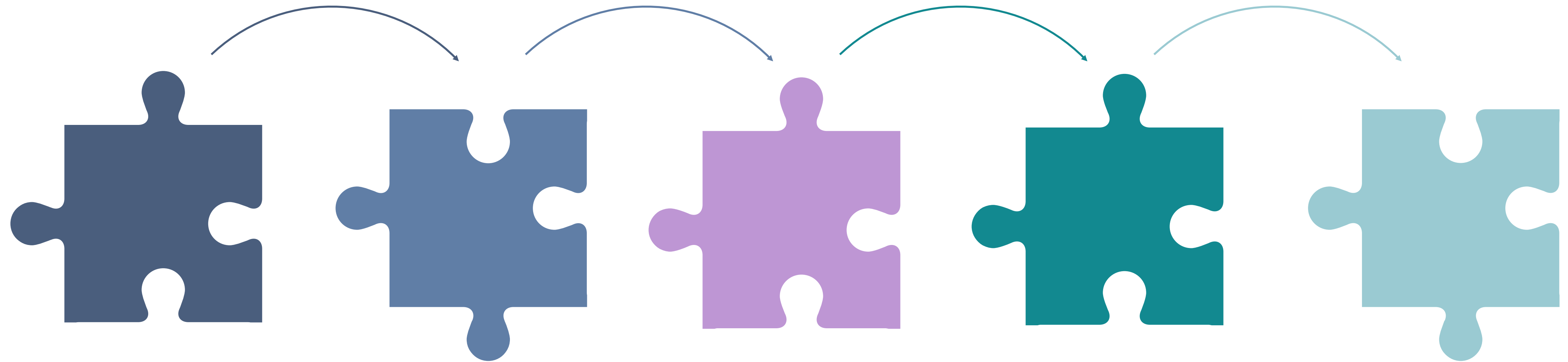
How fast is groundwater or contaminant migrating? Will this be impacted by other effects?

4. Optimized Mitigation

If remedial actions are required, how can they be optimized and become highly effective?

5. Proof and quantification of biodegradation

Determine natural degradation rate.



1. Preferential pathways

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How much contamination is migrating? Is this a relevant mass to be considered a migration risk?

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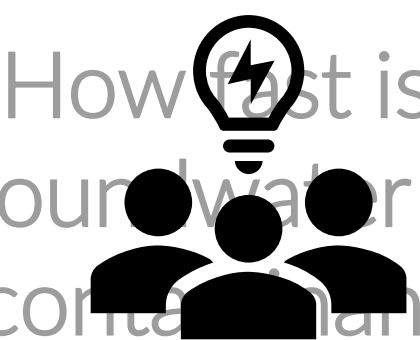
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Determine natural degradation rate.



Iso detect
Umweltmonitoring GmbH



CORNET & REMEDIATION
Belgium
SCIENTIFICS & ENVIRONMENTAL FORENSICS

iFLUX

Steps in Remediation Measures

01

Conditions

02

Proof of
degradation

03

Quantification
of degradation

INCREASING COMPLEXITY OF THE INVESTIGATION

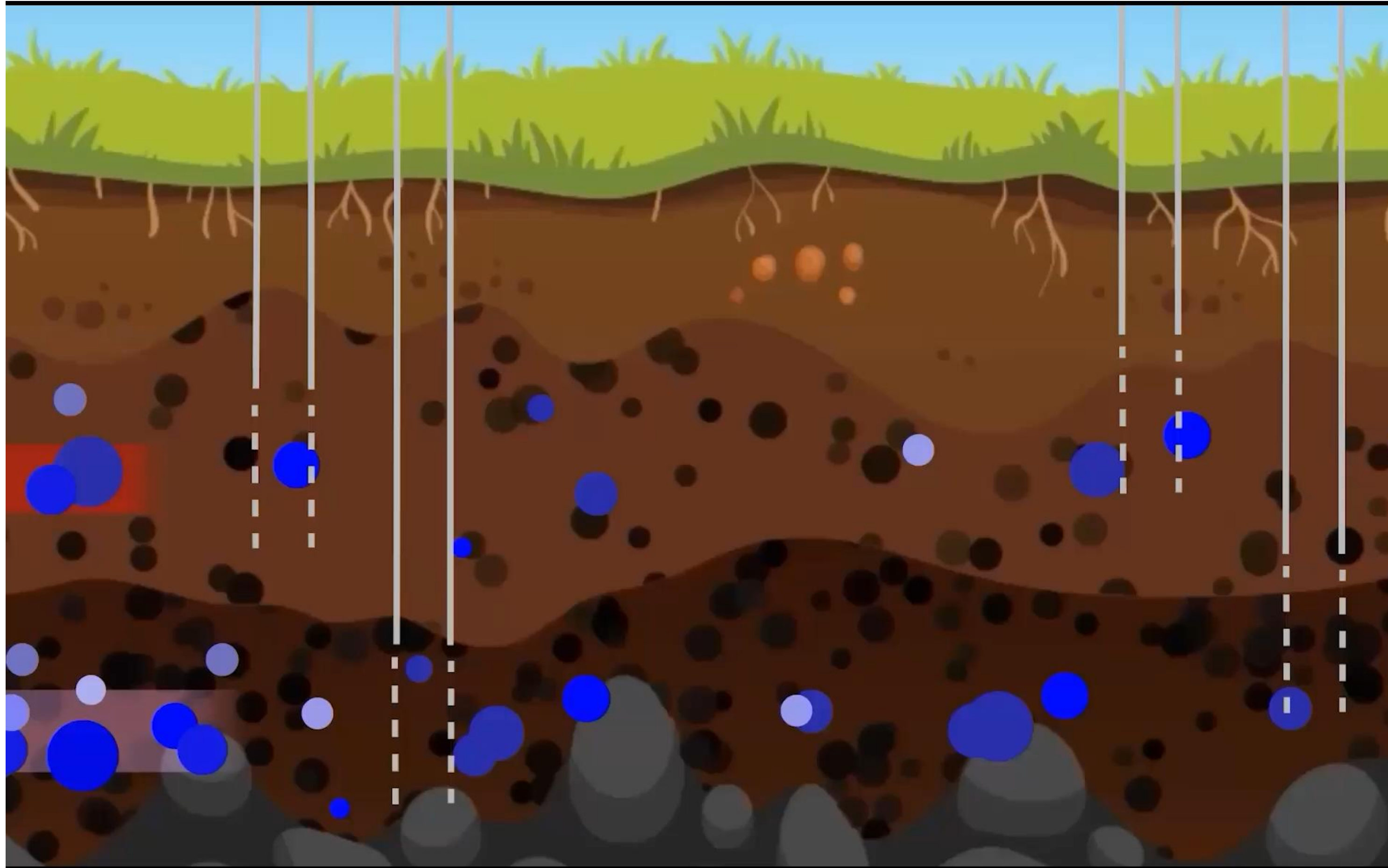
iFLUX
samplers

IsoFLUX samplers

ports
type analysis

The dynamics of contamination

iFLUX Samplers - Contamination & Remediation



iFLUX Samplers

Measuring contamination dynamics for more effective remediation

- reduce remediation costs by up to 40%
- reduce remediation duration by up to 20%
- increase success rates (avoid remediation failure)

“Remediation is extremely costly, so it’s best that our customers do it in a targeted way and get it right the first time.”

Erik Bosmans, project manager iFLUX



Winner of the NICOLE
Innovation Award 2017



Compound-specific Stable Isotope Analysis (CSIA)

...Isotopes are atoms of an element with the same number of protons but different numbers of neutrons

For example Carbon:

^{13}C = 0.96% to 1.16%

^{12}C = 98.84% to 99.04%

$\frac{^{13}\text{C}}{^{12}\text{C}}$ Isotope ratio



measured via isotope ratio mass spectrometry (IRMS)

$$\delta^{13}\text{C} = \frac{\frac{^{13}\text{C}}{^{12}\text{C}}_{\text{Sample}}}{\frac{^{13}\text{C}}{^{12}\text{C}}_{\text{Standard}}} - 1 \quad \text{in } \text{‰}$$

WH

Number of Neutrons

p	1	2	3	4	5	6	7	8
	H	He	Li	Be	B	C	N	O
n	Number of Protons							
0	^1H							
1	^2D	^3He						
2	^3T	^4He	^6Li	^7Be		^8B		
3		^5He	^7Li	^9Be	^{10}B	^{11}C	^{12}N	^{13}O
4		^6He	^8Li	^{10}Be	^{11}B	^{12}C	^{13}N	^{14}O
5			^9Li	^{11}Be	^{12}B	^{13}C	^{14}N	^{15}O
6		^8He	^{10}Li	^{12}Be	^{13}B	^{14}C	^{15}N	^{16}O
7				^{11}Li	^{13}Be	^{14}B	^{15}C	^{16}N
8					^{14}Be	^{15}B	^{16}C	^{17}N
9						^{16}B	^{17}C	^{18}N
10							^{17}B	^{18}C
11								^{19}N
12								^{20}N
13								^{21}N
14								^{22}N
15								^{23}N
16								^{24}N

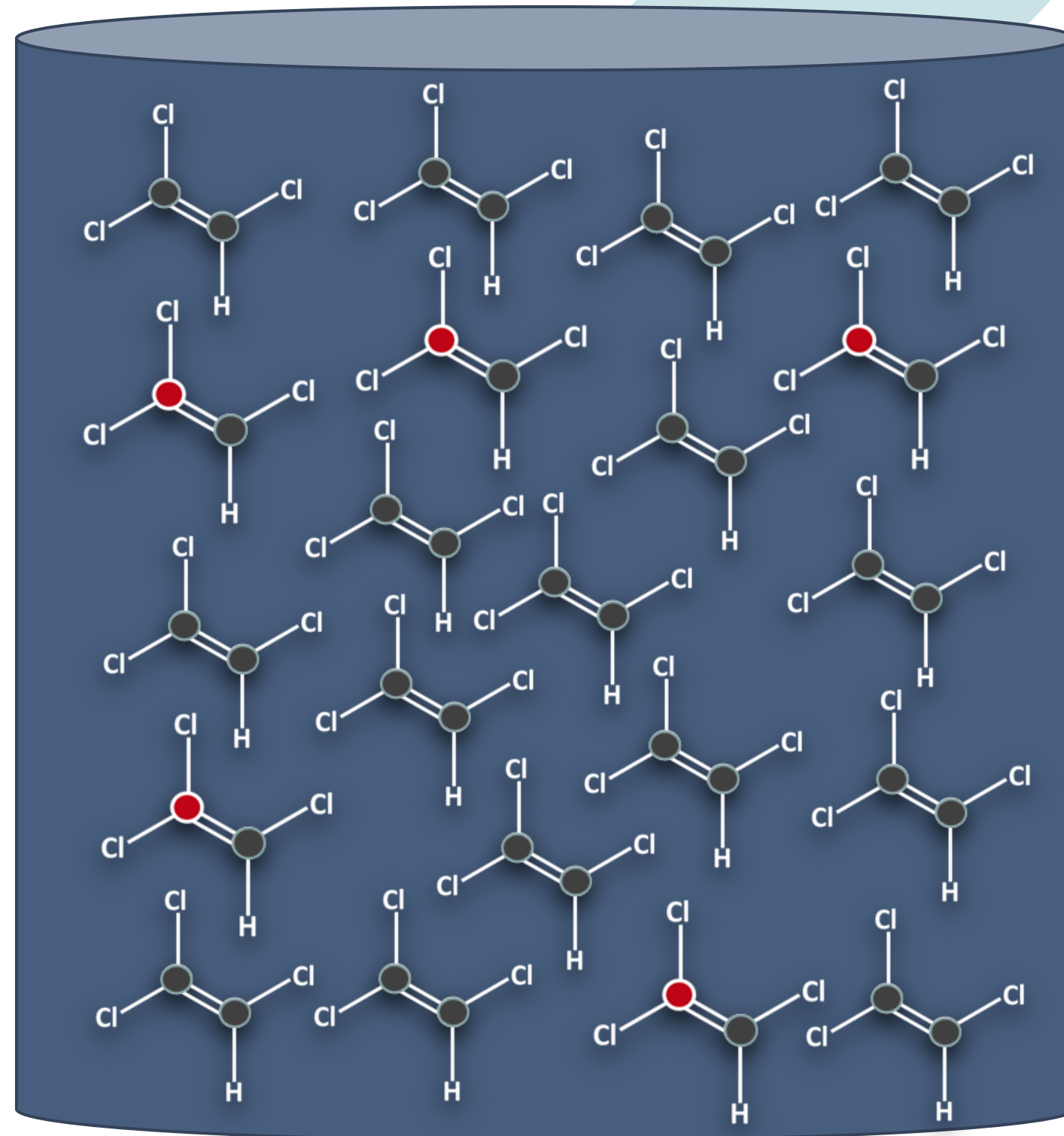
ES?

■ stable isotopes
■ radioactive isotope
■ synthetic

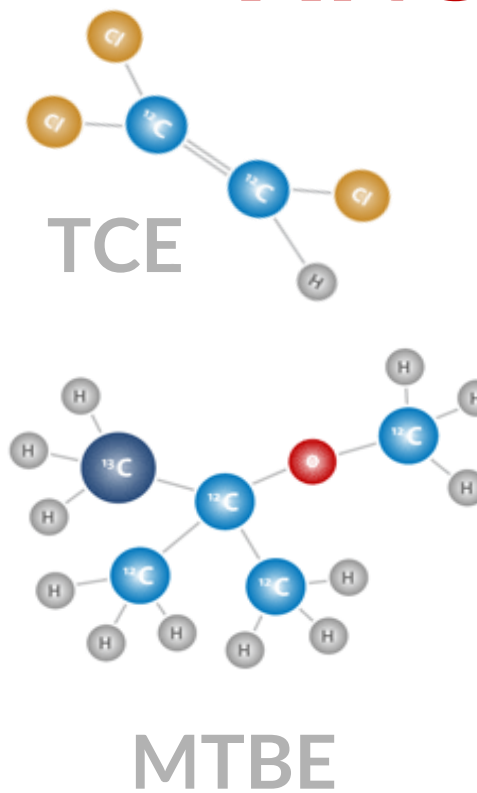
WHY DO I NEED
TO ANALYZE THEM?

pollutant pool.

^{13}C
 ^{12}C



FINGERPRINT – Source Identification



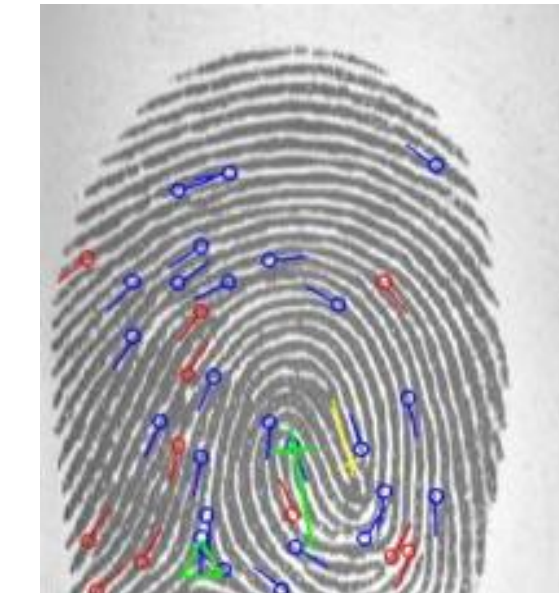
$$\delta^{13}\text{C} = X \text{ ‰ VPDB}$$

$$\delta^{37}\text{Cl} = X \text{ ‰ SMOC}$$

$$\delta^2\text{H} = X \text{ ‰ SMOW}$$

$$\delta^{18}\text{O} = X \text{ ‰ SMOW}$$

$$\delta^{15}\text{N} = X \text{ ‰ AIR}$$

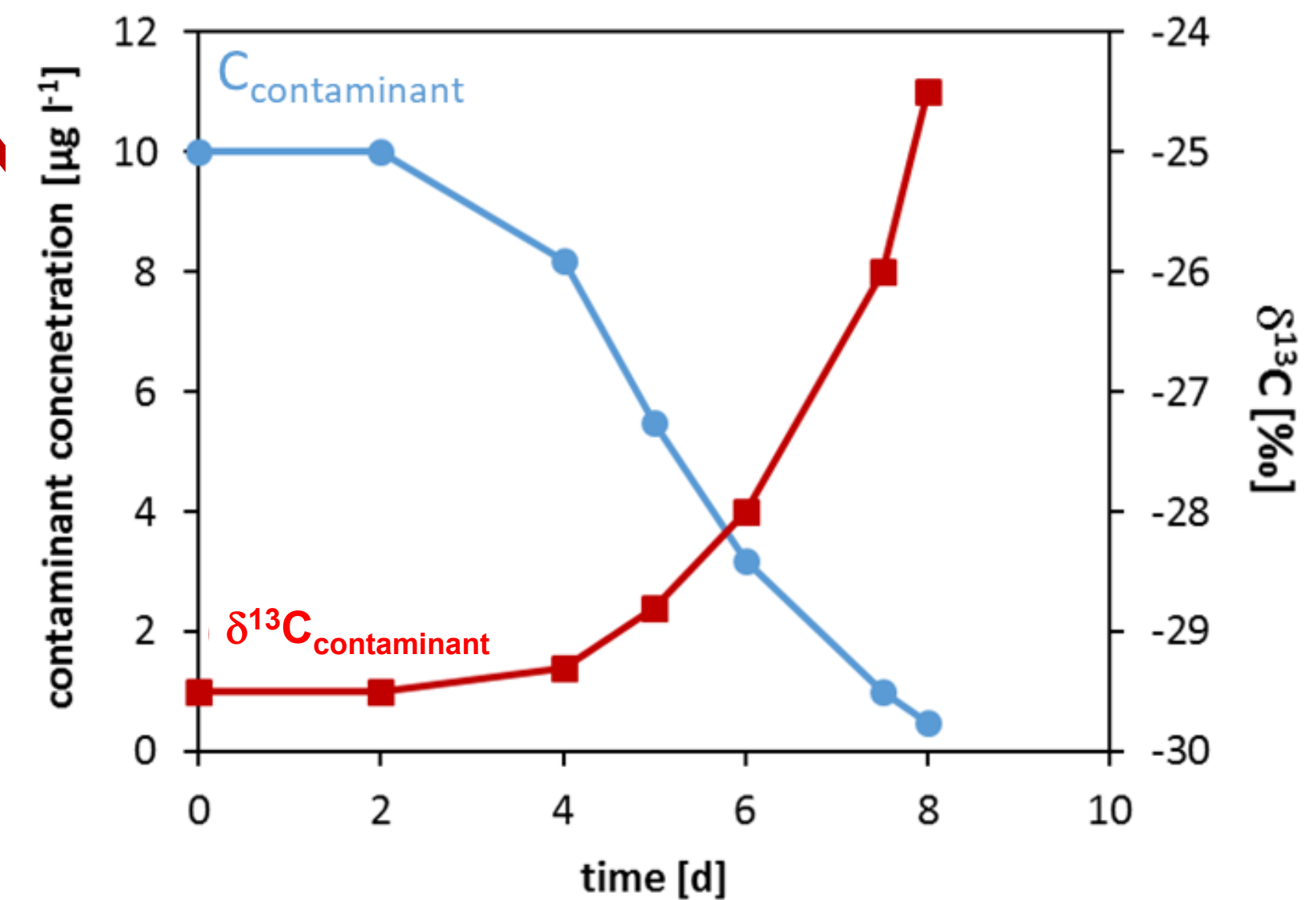


Isotope signature of
the primary

DEGRADATION

^{13}C

^{12}C



In case of degradation, isotope values are getting
more positive

Validated components

iFLUX Samplers - Contamination & Remediation

WATERFLUX Tracer alcohols	CHLOROTOLUENES 2-Chlorotoluene 4-Chlorotoluene	VOLATILE ORGANIC COMPOUNDS SPECIFIC 1,1,1,2-Tetrachloroethane 1,1,2-Trichloro-1,2,2-trifluoroethane 1,1-Dichloroethene 1,1-Dichloropropane 1,1-Dichloropropene 1,2,3-Trichloropropane 1,2-Dibromoethane 1,2-Dichloropropane 1,3-Dichloropropane 2,2-Dichloropropane 2,3-Dichloropropene 2-Chloro-1,3-butadiene 2-Ethyltoluene 3-Chloro-1-propene (allylchloride) 3-Ethyltoluene 4-Ethyltoluene Bromobenzene Bromochloromethane Bromodichloromethane Bromomethane Chloroethane cis-1,3-Dichloropropene Cumene Dibromochloromethane Dibromomethane Diisopropylether ETBE (Ethyl tert-butyl ether) Ethylether Hexachlorobutadiene Iodomethane TAME (Tert-Amyl Methyl Ether) trans-1,3-Dichloropropene Tribromomethane (Bromoform) Trichloromono-fluor- ne	HEAVY METALS Cadmium Chromium Copper Lead Nickel Zinc	PFAS Perfluorobutanoic acid (PFBA) Perfluorohexanoic acid (PFHxA) Perfluoroheptanoic acid (PFHpA) Perfluorooctanoic acid (PFOA) (linear) Perfluorooctanoic acid - sum (PFOATotal) Perfluorononanoic acid (PFNA) Perfluorodecanoic acid (PFDA) Perfluoroundecanoic acid (PFUnDA) Perfluorododecanoic acid (PFDoDA) Perfluorohexadecanoic acid (PFTeDA) Perfluorohexadecanoic acid (PFHxDA) Perfluorobutane acid (PFBS) Perfluorohexane sulfonic acid (PFHxS) (linear) Perfluorohexane sulfonic acid (PFHxSTotal) Perfluoroheptane sulfonic acid (PFHpS) Perfluorooctane sulfonic acid (PFOS) (linear) Perfluorooctane sulfonic acid - sum (PFOSTotal) Perfluorononane sulfonic acid (PFNS) Perfluorodecane sulfonic acid (PFDS) 4:2 Fluorotelomersulfonic acid (4:2 FTS) 6:2 Fluorotelomersulfonic acid (6:2 FTS) 8:2 Fluorotelomersulfonic acid (8:2 FTS) 10:2 Fluorotelomersulfonic acid (10:2 FTS) Perfluorooctane sulfonamide (PFOSA) (linear) Perfluorooctane sulfonamide - sum (PFOSA) (linear) N-ethylperfluorooctane sulfonamide (EtPFOSA) (linear) N-ethylperfluorooctane sulfonamide (EtPFOSATotal) N-methylperfluorooctane-sulfonamidoacetic acid (MePFOSAA)	HEAVY METAL SPECIFIC Mercury Arsenic	NUTRIENT ANIONIC Nitrate-N Sulfate	DIOXANE 1,4-Dioxane	PFAS N-ethylperfluorooctanesulfonamidoacetic acid (EtPFOSAA) 8:2 Fluorotelomerphosphate diester (8:2 DiPAP) 2,3,3,3-tetrafluor-2-(heptafluoropropoxy)propanoic acid (HFPO-DA) Perfluorohexanesulfonamide (PFHxSA) Perfluorotridecanoic acid (PFTeDA) Perfluorooctadecanoic acid (PFODA) Perfluoroundecane sulfonic acid (PFUnDS) Perfluorododecane sulfonic acid (PFDoDS) Perfluorotridecane sulfonic acid (PFTeDS) 6:2 Fluorotelomerphosphate diester (6:2 DiPAP) 6:2/8:2 Fluorotelomerphosphatediester (6:2/8:2 DiPAP) 11-Chloroeicosafluoro-3-oxaundecane-1-sulfonate (11Cl-PF30uDS) (F-53B Minor) 9-Chlorohexadecafluoro-3-oxanonane-1-sulfonate (9Cl-PF30NS) (F-53B Major) Perfluoropentanoic acid (PFPeA)* Perfluoropentane sulfonic acid (PFPeS)* N-methylperfluorooctanesulfonamide (MePFOSA) (linear)* N-methylperfluorooctanesulfonamide (MePFOSATotal)* 4,8-Dioxa-3H-perfluorononocacid (DONA)* Perfluoro-4-ethylcyclohexanesulfonic acid (PFECBS)* Perfluorobutanesulfonamide (PFBSA)* N-methylperfluorobutanesulfonamide (MePFBSA)* N-Methyl-perfluorobutane-sulfonylamidoacetate (MePFBSAA)* 2H-Perfluoro-2-decanoic acid (FOUEA / 8:2 FTUCA)*
BTEX-N-S-MTBE Benzene Toluene Ethylbenzene O-Xylene M,p-Xylenes Naphthalene Styrene MTBE	TRIMETHYLBENZENES 1,2,3-trimethylbenzene 1,2,4-trimethylbenzene 1,3,5-trimethylbenzene		METAL SPECIFIC Calcium Iron Potassium Magnesium Manganese Sodium					
MINERAL OILS Fraction C-10 - C-12 Fraction C-12 - C-20 Fraction C-20 - C-30 Fraction C-30 - C-40 Mineral oils (GC)	POLYAROMATIC HYDROCARBONS Naphthalene Acenaphthylene Acenaphthene Fluorene Fenanthrene Anthracene Fluoranthene Pyrene Benzo(a)anthracene Chrysene Benzo(b)fluoranthene Benzo(k)fluoranthene Benzo(a)pyrene Dibenzo(ah)anthracene Benzo(ghi)perylene Indeno(123cd)pyrene		NUTRIENT CATIONIC Ammonium - N					
CHLORINATED SOLVENTS Dichloromethane 1,1-Dichloroethane 1,2-Dichloroethane Cis-1,2-dichloroethene Trans-1,2-dichloroethene Trichloromethane Trichloroethene 1,1,1-Trichloroethane 1,1,2-Trichloroethane Tetrachloromethane Tetrachloroethene Vinylchloride	POLYCHLORINATED BIPHENYLS PCB 28 PCB 138 PCB 52 PCB 153 PCB 101 PCB 180 PCB 118		HEAVY METAL SPECIFIC Mercury Arsenic					
CHLOROBENZENES Monochlorobenzene 1,2-Dichlorobenzene 1,3-Dichlorobenzene 1,4-Dichlorobenzene 1,2,3-Trichlorobenzene 1,2,4-Trichlorobenzene 1,3,5-Trichlorobenzene	CHLOROTOLUENES 2-Chlorotoluene 4-Chlorotoluene							

Pesticides

iFLUX

SGS

+160 validated components, including 51 types of PFAS

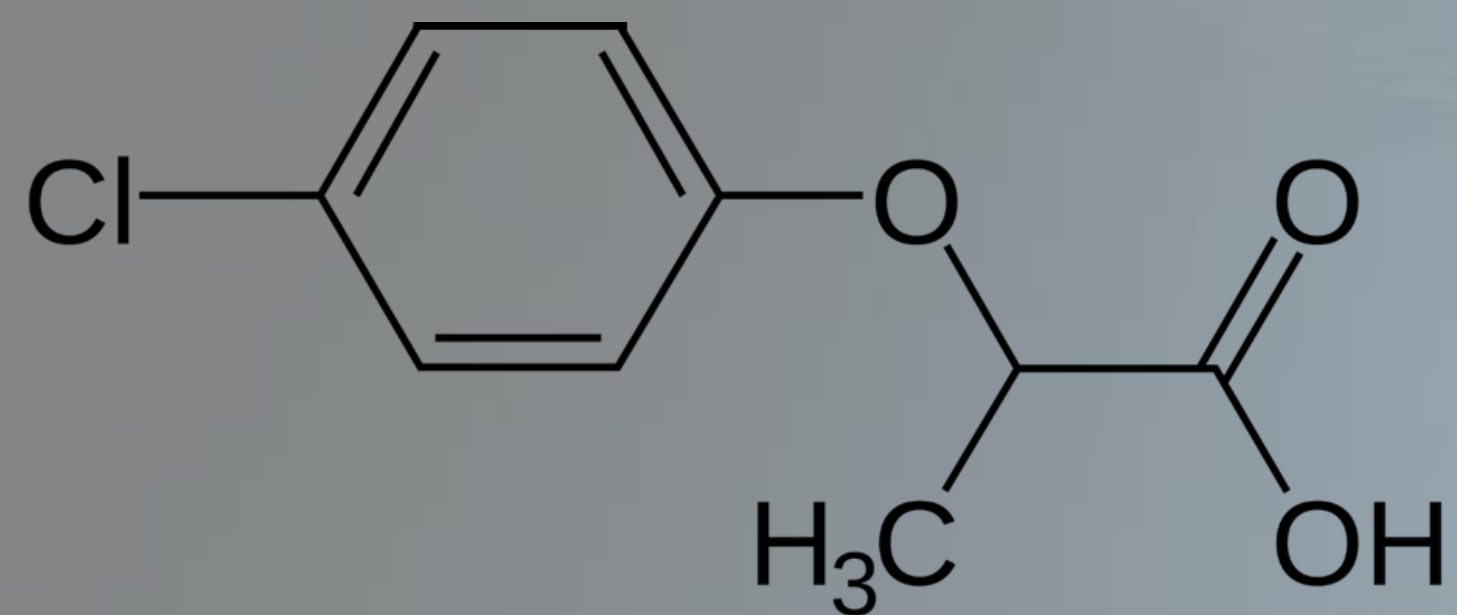
NEW
isoFLUX (δ C¹³/C¹²)
BTEX, chlorethenes

Isodetect
Umweltmonitoring GmbH

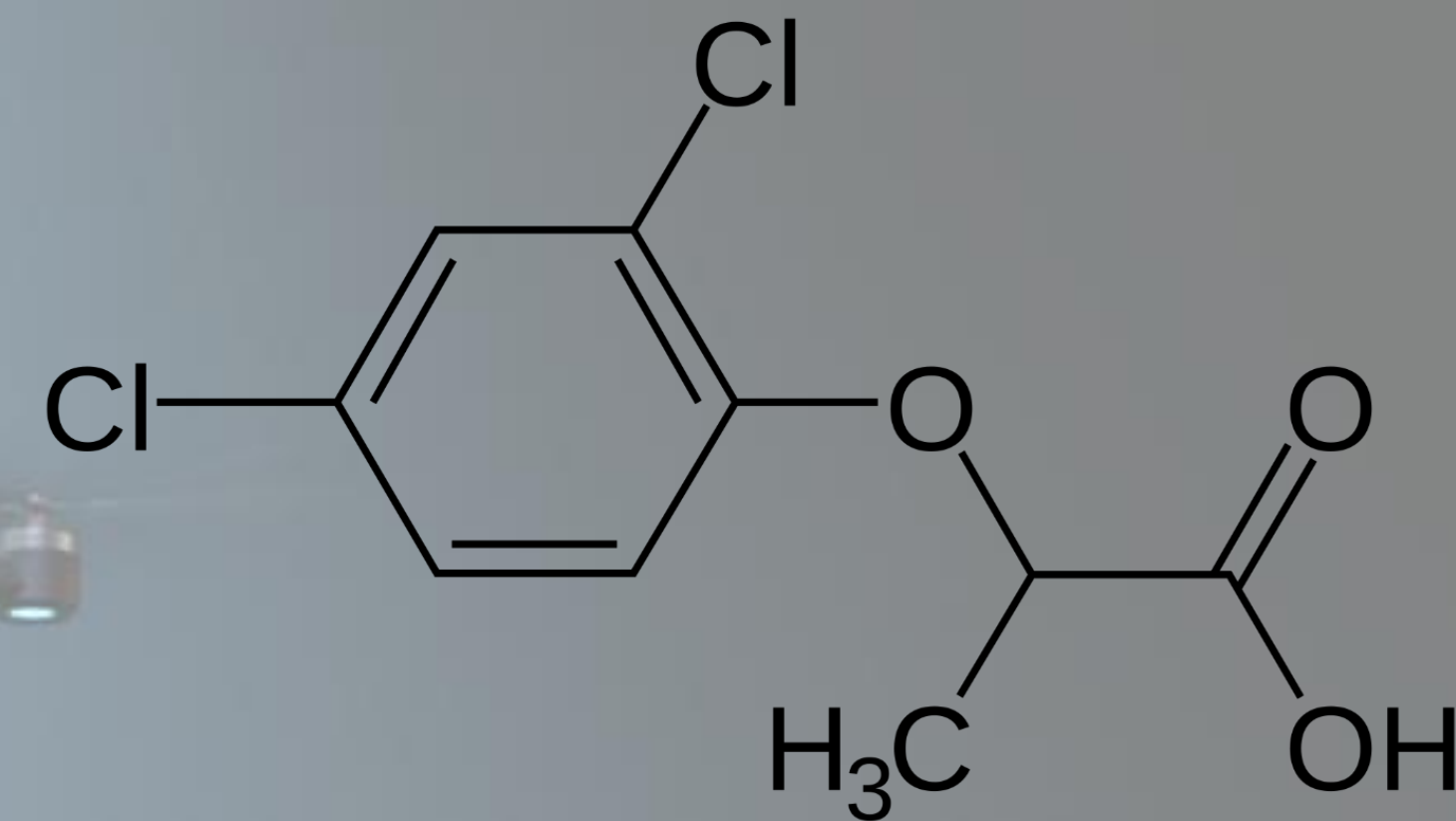
For more details, please visit
www.iFLUX.be

Collaborative third-party labo

SGS



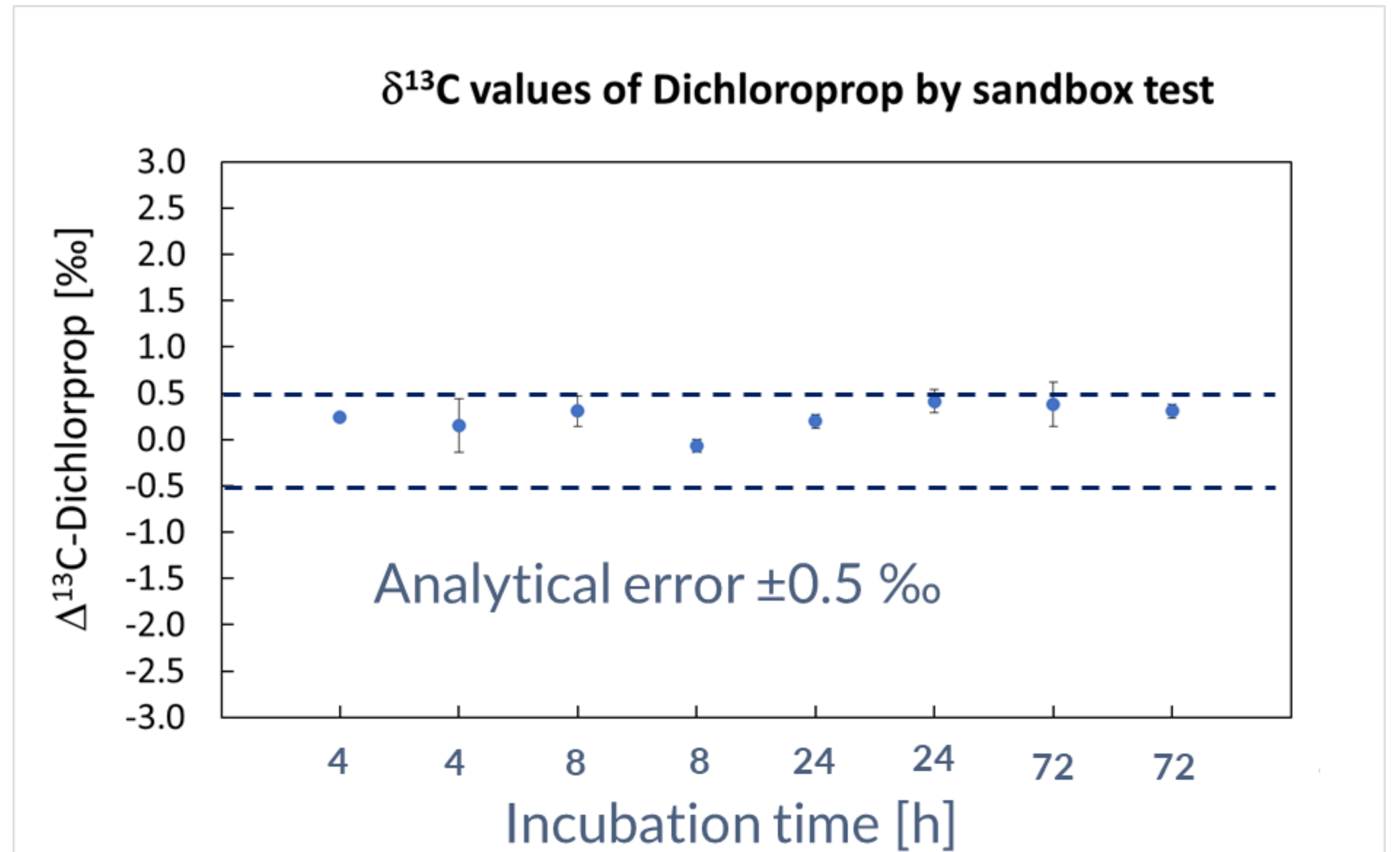
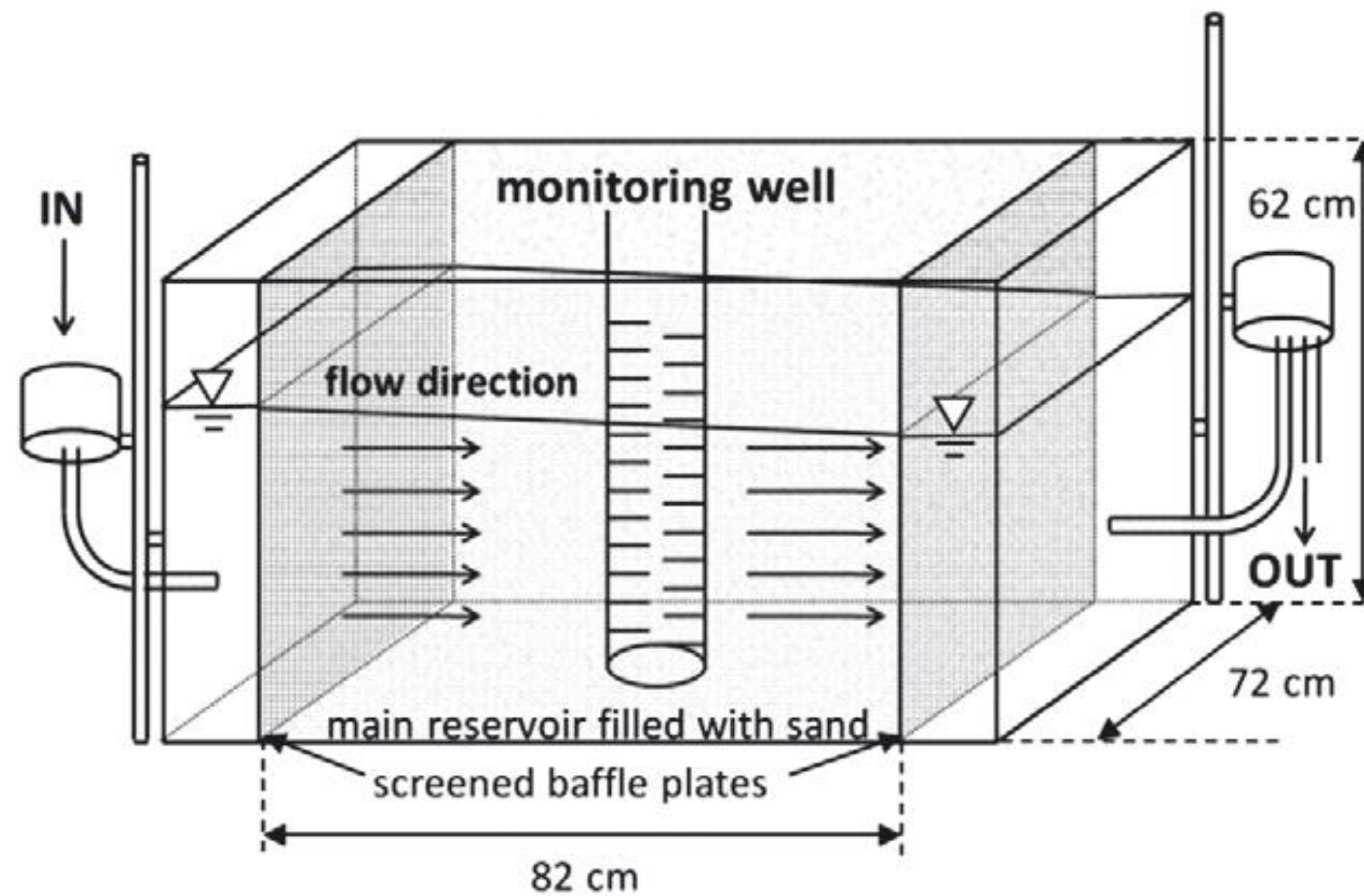
4-CPP
2-(4-Chlorphenoxy)-
Propionic acid



Dichloroprop
2-(2,4-dichlorophenoxy)-
propionic acid

- Flux measurements
- CSIA

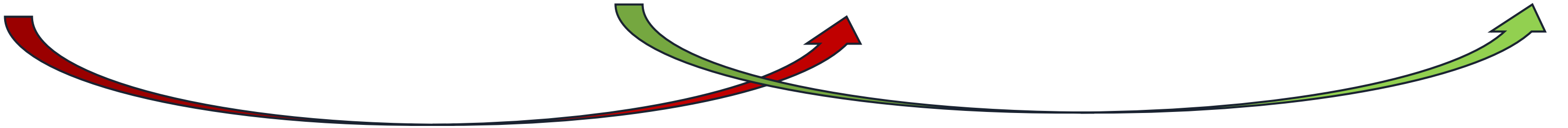
Performance Test in the Lab



Field testing



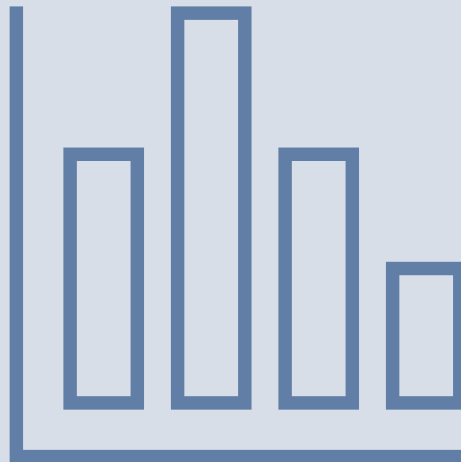
Well B1 primary source	4-CPP	Dichloro-prop	Well B2 secondary source	4-CPP	Dichloro-prop	Well B3 downgradient	4-CPP	Dichloro-prop
Concentration [µg/L]	5.8	5.2	Concentration [µg/L]	22	69	Concentration [µg/L]	0.5	0.12
$\delta^{13}\text{C}$ [‰]	-28.2	-27.3	$\delta^{13}\text{C}$ [‰]	-28.3	-28.9	$\delta^{13}\text{C}$ [‰]	-27.0	-28.2



Secondary source

Proof of degradation of 4-CPP

Isotopes meet mass flux – new evaluation tool for degradation and sources of groundwater pollutants

			
<p>Improve quantification of contaminant flux reduction</p> <p>Comprehensive insights of site conditions over a clearly defined timeframe including water and contaminant flux (& flow direction)</p>	<p>Easy to sample</p> <p>with a small and fixed volume sampler</p>	<p>Level out temporary peaks</p> <p>Sampling passively over a longer period of time avoids temporary peak concentrations</p>	<p>Lower costs</p> <p>Passive collection of analytes in comparison to direct sampling lowers sampling efforts and costs.</p>



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